

## 4-Acetylidene-1-ethyl-2,3,4,5-tetrahydro-1*H*-1,5-benzodiazepin-2-one

Mohammed Sydia Mohamed Said Ould,<sup>a</sup> Bahia Djerrari,<sup>a</sup>  
M'Barek El Abbassi,<sup>a</sup> Jilali Fifani,<sup>a</sup> El-Mokhtar Essassi,<sup>a</sup>  
Brahim El-Bali<sup>b</sup> and Michael Bolte<sup>c\*</sup>

<sup>a</sup>Laboratoire de Chimie Organique Hétérocyclique, Département de Chimie, Faculté des Sciences, Av. Ibn Batouta, Rabat, Morocco, <sup>b</sup>Département de Chimie, Faculté des Sciences Dhar Mehraz, Fés, Morocco, and <sup>c</sup>Institut für Organische Chemie, J.-W.-Goethe-Universität Frankfurt, Marie-Curie-Straße 11, 60439 Frankfurt/Main, Germany

Correspondence e-mail: bolte@chemie.uni-frankfurt.de

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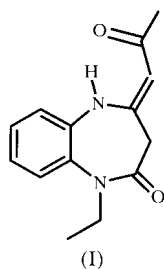
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The title compound, C<sub>14</sub>H<sub>16</sub>N<sub>2</sub>O<sub>2</sub>, contains a diazepine ring, which appears in a boat conformation. An intramolecular hydrogen bond is formed between the NH group of the diazepine ring and a carbonyl O atom of one of the side chains.

### Comment

As part of our research on the use of 4-hydroxy-6-methylpyran-2-one in heterocyclic synthesis (El Abbassi *et al.*, 1987, 1997; Essassi *et al.*, 1987), we have reinspected the condensation of *o*-phenylenediamine with  $\gamma$ -pyrone with the aim of confirming the structure of the obtained product. In a previous work (El Abbassi *et al.*, 1987), we have shown that the reaction of *o*-phenylenediamine with  $\gamma$ -pyrone leads to 4-acetylidene-1,5-benzodiazepin-2-one. In order to confirm the structure of the reaction product, we have converted 4-acetylidene-1,5-benzodiazepin-2-one with ethyl bromide to afford the title compound, (I), and carried out an X-ray structure analysis.



The geometry of the title compound shows no unusual features. The diazepine ring shows a boat conformation. The seven-membered ring can be described as being composed of three planes: a bow plane (C2, C3 and C4), a central plane

(N1, C2, C4 and N5; r.m.s. deviation 0.019 Å) and a stern plane (N1, C11, C6 and N5; r.m.s. deviation 0.013 Å). The dihedral angle between central and bow planes is 58.70 (8)°, and there is an angle of 36.31 (5)° between central and stern planes.

The title compound shows an intramolecular hydrogen bond between the NH group of the diazepine ring and a carbonyl O atom of one of the side chains. Furthermore, the other carbonyl O atom shows three short C—H...O contacts.

### Experimental

To a solution of 4-acetylidene-1,5-benzodiazepin-2-one (1 mmol), in benzene (60 ml), tetra-*n*-butylammonium bromide (20 mmol) in NaOH (30 ml) was added as an alkylating agent. After dilution and decantation, the organic phase was washed with a hydrochloric acid solution (10%) and dried. The benzene was evaporated under reduced pressure and the residue was chromatographed on a silicate column with hexane as eluant. Analysis measured (calculated): C 68.83 (68.85), H 6.52 (6.56), N 11.43% (11.47%).

### Crystal data

C <sub>14</sub> H <sub>16</sub> N <sub>2</sub> O <sub>2</sub>	$D_x = 1.249 \text{ Mg m}^{-3}$
$M_r = 244.29$	Mo $K\alpha$ radiation
Monoclinic, $P2_1/n$	Cell parameters from 8192 reflections
$a = 8.730 (1) \text{ \AA}$	$\theta = 1-25^\circ$
$b = 10.052 (1) \text{ \AA}$	$\mu = 0.085 \text{ mm}^{-1}$
$c = 14.998 (2) \text{ \AA}$	$T = 173 (2) \text{ K}$
$\beta = 99.36 (1)^\circ$	Block, colourless
$V = 1298.6 (3) \text{ \AA}^3$	$0.60 \times 0.40 \times 0.30 \text{ mm}$
$Z = 4$	

### Data collection

Siemens CCD three-circle diffractometer	$R_{\text{int}} = 0.034$
$\omega$ scans	$\theta_{\text{max}} = 28.70^\circ$
Absorption correction: empirical ( <i>SADABS</i> ; Sheldrick, 1996)	$h = -11 \rightarrow 11$
$T_{\text{min}} = 0.951$ , $T_{\text{max}} = 0.975$	$k = -13 \rightarrow 13$
28 486 measured reflections	$l = -20 \rightarrow 20$
3350 independent reflections	8192 standard reflections
2865 reflections with $I > 2\sigma(I)$	frequency: 1080 min
	intensity decay: none

### Refinement

Refinement on $F^2$	$w = 1/[\sigma^2(F_o^2) + (0.0635P)^2 + 0.2665P]$
$R[F^2 > 2\sigma(F^2)] = 0.039$	where $P = (F_o^2 + 2F_c^2)/3$
$wR(F^2) = 0.113$	$(\Delta/\sigma)_{\text{max}} < 0.001$
$S = 1.068$	$\Delta\rho_{\text{max}} = 0.29 \text{ e \AA}^{-3}$
3350 reflections	$\Delta\rho_{\text{min}} = -0.26 \text{ e \AA}^{-3}$
164 parameters	
H-atom parameters constrained	

**Table 1**

Hydrogen-bonding geometry (Å, °).

$D-H \cdots A$	$D-H$	$H \cdots A$	$D \cdots A$	$D-H \cdots A$
C10—H10...O41 <sup>i</sup>	0.95	2.52	3.3367 (13)	144
C21—H21...O41 <sup>ii</sup>	0.95	2.58	3.4457 (14)	152
C3—H3A...O41 <sup>ii</sup>	0.99	2.66	3.5948 (14)	157
N1—H1...O22	0.88	1.98	2.6640 (12)	133

Symmetry codes: (i)  $1 + x, y, z$ ; (ii)  $-x, 1 - y, 1 - z$ .

All H atoms were initially located by difference Fourier synthesis. Subsequently their positions were idealized and constrained to ride on their parent atoms with C—H(aromatic) = 0.95, C—H(secondary) = 0.99, C—H(methyl) = 0.98 or N—H = 0.88 Å, and fixed individual displacement parameters [ $U(\text{H}) = 1.5U_{\text{eq}}(\text{methyl C})$ ].

$U(H) = 1.2U_{eq}(C)$  or  $U(H) = 1.2U_{eq}(N)$ ]. The C23 methyl group was allowed to rotate about its local threefold axis.

Data collection: *SMART* (Siemens, 1995); cell refinement: *SMART*; data reduction: *SAINT* (Siemens, 1995); program(s) used to solve structure: *SHELXS97* (Sheldrick, 1990); program(s) used to refine structure: *SHELXL97* (Sheldrick, 1997).

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